

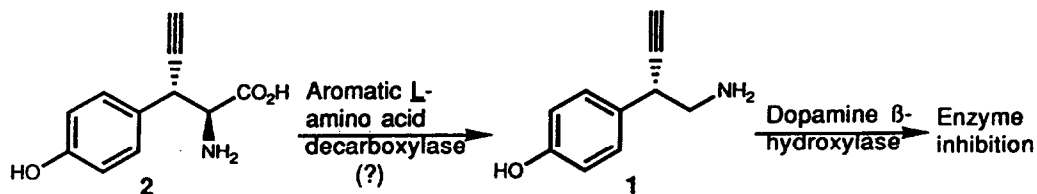
ENANTIOSELECTIVE SYNTHESIS OF (+)-(2S,3S)-3-ETHYNYLTYROSINE

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Summary: Homochiral ethynyl tyrosine 2, designed as a potential dopamine β -hydroxylase irreversible inhibitor prodrug, has been synthesised asymmetrically in 15-steps, using the alkylation and azidation of acylated oxazolidones to establish relative and absolute stereochemistry.

2S-Ethynyltyramine 1 is a high affinity mechanism-based inhibitor of dopamine β -hydroxylase (DBH).² As a means of enhancing the *in vivo* pharmacological profile of 1, ethynyltyrosine 2 was designed as a potential dual enzyme-activated irreversible inhibitor³ of DBH. Active amino acid transport of 2 into cells and the subsequent action of aromatic L-amino acid decarboxylase (AADC)⁴ was envisaged to deliver 1 to target organelles.² In this Letter we describe the enantioselective synthesis of (+)-2. The stereocenters at C(3) and C(2) are introduced via asymmetric alkylation and azidation reactions, using Evans' chiral oxazolidones as stereocontrollers.^{5,6}

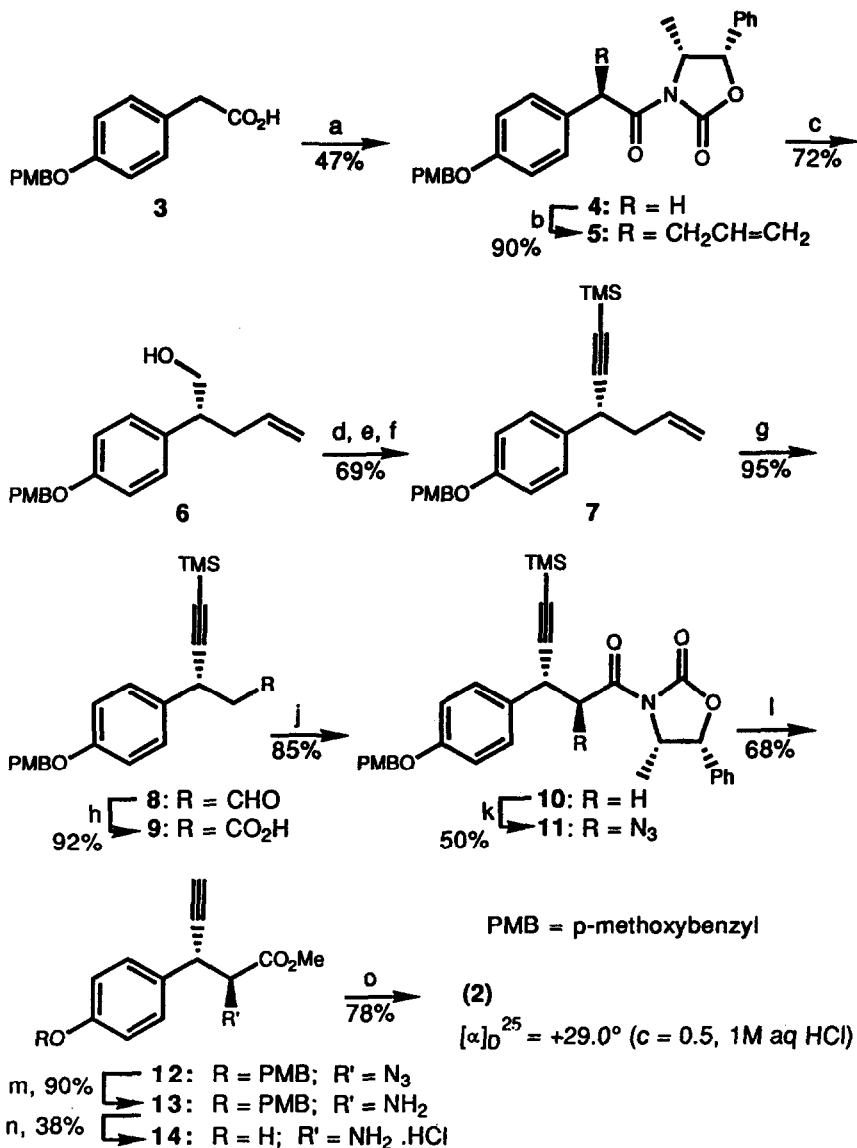


The synthesis of (+)-2 was initiated upon coupling aryl acetic acid 3⁷ with (+)-(4R,5S)-4-methyl-5-phenyloxazolidone to give the N-acylated derivative 4^{5a,8} (Scheme 1). Alkylation of the sodium enolate of 4 with freshly distilled allyl

bromide (1.1 equiv. NaHMDS, $-78\text{ }^{\circ}\text{C}$, 15 min; 5 equiv. $\text{CH}_2=\text{CHCH}_2\text{Br}$, $-50\text{ }^{\circ}\text{C}$, 1h) furnished oxazolidone 5 in 90% yield.^{5b} The alkylated product contained ca. 8% (200 MHz ^1H NMR analysis) of the unwanted diastereomer which was inseparable by chromatography and was conveniently removed later in the synthesis. Subsequent removal of the chiral auxiliary (LAH, THF, $0\text{ }^{\circ}\text{C}$, 30 min) gave alcohol 6 which was converted to the enyne 7 using Corey's procedure:⁹ 1) Swern oxidation, 2) Wittig condensation with $\text{Ph}_3\text{P}=\text{CBr}_2$ (2 equiv.) and 3) treatment with $t\text{-BuLi}$ (2 equiv.) and Me_3SiCl (1.5 equiv., 69% overall). Selective ozonolysis of 7 in CH_2Cl_2 ($-78\text{ }^{\circ}\text{C}$) using Sudan 7B as an indicator¹⁰ and reductive work up with Ph_3P produced aldehyde 8. Further oxidation of 8 with NaClO_2 in aqueous $t\text{-BuOH}$ ¹¹ gave the known acid 9 ($[\alpha]_{\text{D}}^{25} = -16.2^{\circ}$ (c 1.5, DMF) (lit.² $[\alpha]_{\text{D}}^{25} = -19.4^{\circ}$ (c 1.5, DMF)). The magnitude of $[\alpha]_{\text{D}}^{25}$ for synthetic 9 implied an ee of 84% indicating that ca. 5% racemization at C(3) occurred during the 6 \rightarrow 7 transformation while the (-) sign of $[\alpha]_{\text{D}}^{25}$ confirmed the stereochemistry at C(3) as S .²

Introduction of the amine functionality present in 2 was accomplished via an azidation/reduction sequence. Accordingly, oxazolidone 10 derived from 9 and (-)-(4S,5R)-4-methyl-5-phenyloxazolidone was subjected to electrophilic azidation as described by Evans.^{5c} Thus a solution of 10 (0.25 M in THF) was cooled to $-100\text{ }^{\circ}\text{C}$ and treated with KHMDS (1.1 equiv., 30 min) followed by the addition of a precooled ($-78\text{ }^{\circ}\text{C}$) solution of 2,4,6-triisopropylbenzenesulfonyl azide (3 equiv. THF). The reaction was stirred at $-78\text{ }^{\circ}\text{C}$ for 5 min and then quenched with AcOH (5 equiv., $-78\text{ }^{\circ}\text{C} \rightarrow 30\text{ }^{\circ}\text{C}$, 1 h). Extractive workup and chromatography afforded azide 11 in 50% yield (80% de). The use of alternative bases or sulfonyl azides did not improve the yield. Following conversion of 11 to azido ester 12 (three steps: 1) peroxide mediated hydrolysis of the acyl oxazolidone,^{5c,d} 2) removal of the Me_3Si protecting group and 3) esterification), azide reduction was accomplished efficiently using Ph_3P (2 equiv. in aqueous THF, reflux, 3 h),¹² procuring amine 13. Tyrosine ester 14 ($[\alpha]_{\text{D}}^{25} = +30.4^{\circ}$ (c 1.1, H_2O)) was obtained as a single diastereomer in 30% yield upon removal of the PMB protecting group from 13 (1 M methanolic HCl, $25\text{ }^{\circ}\text{C}$, 12 h) and two recrystallizations of the phenol from $\text{MeOH}-\text{Et}_2\text{O}$ (1:1).¹³ Final hydrolysis of 14 (3 M aqueous HCl at $60\text{ }^{\circ}\text{C}$, 24 h) completed the preparation of enantiomerically pure (+)-2, $[\alpha]_{\text{D}}^{25} = +29.0^{\circ}$ (c 0.5, 1 M aq HCl).

Scheme 1



Reagents and conditions: a) i) t -BuCOCl, Et₃N, THF, 0°C; ii) Li salt of (4R,5S)-4-methyl-5-phenyl-oxazolidone, 0°C; b) i) 1.1 equiv. NaHMDS, 30 min, THF, -78°C; ii) 5 equiv. allyl bromide, -50°C; c) LAH, THF, 0°C; d) i) (COCl)₂, DMSO, CH₂Cl₂, -78°C ii) Et₃N; e) 2 equiv. CBr₄, 4 equiv. PPh₃, CH₂Cl₂; f) i) 2.1 equiv. t -BuLi, -78°C, 30min, to 20°C, 1h, ii) 1.2 equiv. TMSCl; g) i) O₃, sudan red 7B, CH₂Cl₂, -78°C, ii) 2 equiv. PPh₃; h) 9 equiv. NaClO₂, NaH₂PO₄, aq t -BuOH, 2-methyl-2-butene; j) i) t -BuCOCl, Et₃N, THF, 0°C, ii) Li salt of (4S,5R)-4-methyl-5-phenyloxazolidone, 0°C; k) i) 1.05 equiv. KHMDS, -100°C, THF, 30 min, ii) 3 equiv. 2,4,6-tri-isopropylbenzenesulphonyl azide, -78°C, 5 min; iii) 5 equiv. AcOH, -78°C to 30°C; l) i) 2 equiv. LiOH, 4 equiv. H₂O₂, aq THF, 0°C; ii) K₂CO₃, aq MeOH; iii) 1.2 equiv. K₂CO₃, 1.2 equiv. MeI, DMF; m) 2 equiv. PPh₃, aq THF, reflux; n) HCl, MeOH, then recrystallisation (MeOH: Et₂O); o) 25% aq HCl, 60°C, then pyridine.

REFERENCES AND NOTES

1. Present address: Sterling Drug Inc., Malvern, PA, 19355.
2. DeWolf, W.E., Jr.; Chambers, P.A.; Southan, C.; Saunders, D.; Kruse, L.I. Biochemistry, 1989, **28**, 3833 and ref. therein.
3. Dual enzyme-activated irreversible inhibitors of monoamine oxidase have recently been described: McDonald, I.A.; Lacoste, J.M.; Bey, P.; Wagner, J.; Zreika, M.; Palfreyman, M.G. Bioorg. Chem., 1986, **14**, 103.
4. a) Lovenberg, W.; Weissbach, H.; Udenfriend, S. J. Biol. Chem., 1962, **237**, 89. b) Jung, M.J. Bioorg. Chem., 1986, **14**, 429. c) See the discussion in ref 3 above.
5. a) Evans, D.A. Aldrichim. Acta., 1982, **15**, 23. b) Evans, D.A.; Weber, A.E. J. Am. Chem. Soc., 1986, **108**, 6757. c) Evans, D.A.; Britton, T.C. J. Am. Chem. Soc., 1987, **109**, 6881. d) Evans, D.A.; Britton, T.C.; Ellman, J.A. Tetrahedron Lett., 1987, **28**, 6141.
6. See also Dharanipragada, R.; Nicolas, E.; Toth, G.; Hruby, V.J. Tetrahedron Lett., 1989, **30**, 6841, 6845.
7. Kuchar, M.; Brunova, B.; Rejholec, V.; Grimova, J.; Nemecek, O. Collect. Czech. Chem. Commun., 1977, **42**, 1723.
8. All new compounds exhibited physical and spectroscopic properties consistent with their structure. 7: $^1\text{H NMR}$ δ (CDCl_3 , 200 MHz) 7.37-7.23 (m, 4H), 6.94-6.89 (m, 4H), 5.84 (m, 1H, $\text{CH}=\text{CH}_2$), 5.07-4.99 (m, 2H, $\text{CH}=\text{CH}_2$), 4.97 (s, 2H, $p\text{-MeOC}_6\text{H}_4\text{CH}_2$), 3.81 (s, 3H, $p\text{-MeOC}_6\text{H}_4$), 3.66 (t, 1H, $\text{CHC}=\text{CTMS}$, $J=7.0$ Hz), 2.45 (m, 2H, $\text{CH}_2\text{CH}=\text{CH}_2$), 0.17 (s, 9H, TMS). 13: $^1\text{H NMR}$ δ (CDCl_3 , 200 MHz) 7.37-7.21 (m, 4H), 6.95-6.89 (m, 4H), 4.97 (s, 2H, $p\text{-MeOC}_6\text{H}_4\text{CH}_2$), 4.08 (dd, 1H, $\text{CHC}=\text{CH}$, $J=5.3$, 2.5 Hz), 3.83 (d, 1H, CHCO_2Me , $J=5.3$ Hz), 3.82 (s, 3H, $p\text{-MeOC}_6\text{H}_4$), 3.69 (s, 3H, CO_2Me), 2.36 (d, 1H, $\text{C}=\text{CH}$, $J=2.5$ Hz). 2: $^1\text{H NMR}$ (D_2O , 250 MHz) 7.38-7.35 (m, 2H), 6.94-6.90 (m, 2H), 4.59 (dd, 1H, $\text{CHC}=\text{CH}$, $J=4.3$, 2.6 Hz), 4.40 (d, 1H, CHCO_2H , $J=4.3$ Hz), 2.96 (d, 1H, $\text{C}=\text{CH}$, $J=2.6$ Hz).
9. Corey, E.J.; Fuchs, P.L. Tetrahedron Lett., 1972, 3769.
10. Veysoglu, T.; Mitscher, L.A.; Swayze, J.K. Synthesis, 1980, 807.
11. Bal, B.S.; Childers, W.E., Jr.; Pinnick, H.W. Tetrahedron, 1981, **37**, 2091 and ref. therein.
12. a) Vaultier, M.; Knouzi, N.; Carrie, R. Tetrahedron Lett., 1983, **24**, 763. b) Direct reduction of the azido acid precursor of 12 was found to be much slower and poorer yielding.
13. Direct inspection of ester 14 and the corresponding bis-Mosher's acid derivative by 250 MHz $^1\text{H NMR}$ revealed 14 to be stereochemically homogeneous.

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